This article was downloaded by: On: *27 January 2011* Access details: *Access Details: Free Access* Publisher *Taylor & Francis* Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Organic Preparations and Procedures International Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t902189982

A SIMPLE, INEXPENSIVE SYSTEM FOR PERFORMING HIGH PRESSURE REACTIONS

Philip Deshong^a; C. Michael Dicken^a; Joseph J. Perez^a; Robert M. Shoff^b ^a Department of Chemistry, The Pennsylvania State University, University Park, PA ^b Tem-Pres Division, Leco Corporation, Bellefonte, PA

To cite this Article Deshong, Philip , Dicken, C. Michael , Perez, Joseph J. and Shoff, Robert M.(1982) 'A SIMPLE, INEXPENSIVE SYSTEM FOR PERFORMING HIGH PRESSURE REACTIONS', Organic Preparations and Procedures International, 14: 5, 369 — 372 **To link to this Article: DOI:** 10.1080/00304948209354935

URL: http://dx.doi.org/10.1080/00304948209354935

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

kopf, Ber., 42, 617 (1909)].

- Tautomerization of a nitrolic acid into a nitronic acid is apparently unknown.
- P. A. S. Smith and J. E. Robertson, J. Am. Chem. Soc., <u>84</u>, 1197 (1962).
- P. A. S. Smith, "Open-chain Nitrogen Compounds", Vol. II,
 p. 431, W. A. Benjamin, Inc., New York, 1966.

A SIMPLE, INEXPENSIVE SYSTEM FOR PERFORMING

HIGH PRESSURE REACTIONS

Submitted by Philip DeShong,*[†] C. Michael Dicken,[†] Joseph J. (2/26/82) Perez,[†] and Robert M. Shoff^{††}

^TDepartment of Chemistry The Pennsylvania State University University Park, PA 16802

⁺⁺Leco Corporation, Tem-Pres Division Bellefonte, PA 16823

A pressure system capable of maintaining reaction volumes at pressures of 1-8 Kbar was required because previously reported systems¹ have a variety of flaws:

 They required elaborate pumping apparatus for attaining the required pressure and are therefore, expensive.

2. Large reaction volumes could not be employed.

 The system utilized a gas-pressurized reaction vessel, thus introducing safety considerations.

We now report a high pressure system which is mechanically simple, inexpensive to build, safe, and which can be used with

369



Figure 1. Schematic of High Pressure System



glass plug

heat shrinkable teflon tubing



synthetically significant volumes of reactants.

A schematic of the system is presented in Fig. 1 and consists of a reinforced, steel chamber which is pressurized by a hydraulically-driven piston using unsupported area seals. A smaller hydraulic device drives the piston (Castor oil is employed as the high pressure medium). Reaction pressures are measured indirectly by measuring the pressure generated in the smaller hydraulic device which has been calibrated. This indirect method alleviates the need for an expensive, ultra high pressure gauge.² An additional problem encountered concerned the separation of the chemical reactants from the pressurizing fluid. A variety of reaction containers were investigated such as metal, teflon, polymeric substances and were found to be deficient. The container does not have to withstand tremendous pressures since the surrounding environment is isobaric, but must be capable of contracting and expanding during the reaction. Finally the container has to be inert to an assortment of organic reagents. It was found that a simple container can be made from a length of heat-shrinkable Teflon tubing (Ace Glass Co.) of an appropriate diameter and length to accomodate the volume of reactants, plugged at each end with a small section of solid glass rod (Fig. 2). Care is taken to remove gas bubbles after the reaction mixture has been introduced. This system has never failed even under conditions as extreme as 75°/8 Kbar/48 hrs.

REFERENCES

M. Tkacz and B. Baranowski, Rocz. Chem., <u>50</u>, 2159 (1976);
 W. LeNoble, Prog. Phys. Org. Chem., <u>5</u>, 207 (1967); F.

371

Fleischmann, E. Conze, D. Stranks and H. Kelm, Rev. Sci. Instrum., <u>45</u>, 1427 (1974); R. Greiger and C. Eckert AIChE J., <u>16</u>, 766 (1970). K. Brower, J. Am. Chem. Soc., <u>90</u>, 5401 (1968); T. Moriyoshi; Bull. Chem. Soc. Japan, <u>44</u>, 2582 (1971); H. Yamada, Chem. Lett., 747 (1972). A. Jost, Ber. Bunsenges. Phys. Chem., <u>79</u>, 850 (1975); A. Yu, M. Waissbluth and R. Grieger, Rev. Sci. Instrum., <u>44</u>, 1390 (1973); J. Jonas, ibid, <u>43</u>, 643 (1972).

2. The apparatus can be obtained commercially from Leco Corporation, Tem-Pres Division, Bellefonte, PA 16823.